Roll No.

D-978

M. Sc. (Fourth Semester) (Main/ATKT) EXAMINATION, May-June, 2020

CHEMISTRY

Paper No. CH-19

(Instrumental Methods of Analysis)

Time: Three Hours [Maximum Marks: 80

Note: Attempt all Sections as directed.

Section—A 1 each

(Objective/Multiple Choice Questions)

Note: Attempt all questions.

Choose the correct answer:

- 1. Which of the following is strong basic anion exchanger?
 - (a) $Aryl-SO_3^-H^+$
 - (b) $R-SO_3^-H^+$
 - (c) R-COO-Na+
 - (d) Aryl $-CH_2N(CH_3)_3^+Cl^-$

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2. The softening of hard water is performed through which of the following chromatography?

- (a) Ion exchange chromatography
- (b) Supercritical chromatography
- (c) Gas chromatography
- (d) High performance liquid chromatography
- 3. Which of the chromatography is good for determination of high molecular weight of polymers?
 - (a) Paper chromatography
 - (b) Thin layer chromatography
 - (c) Size exclusion chromatography
 - (d) Ion exchange chromatography
- 4. Electrochromatography is combination of which two analytical techniques?
 - (a) GC-CE
 - (b) HPLC-CE
 - (c) TLC-CE
 - (d) HPTLC-CE
- 5. The column of capillary electrophoresis is coated with which of the following components?
 - (a) Silica
 - (b) Alumina
 - (c) Dextran
 - (d) Cellulose

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- 6. Which of the following gas can be used as a carrier gas in supercritical fluid chromatography?
 - (a) NO₂
 - (b) CO
 - (c) SO_2
 - (d) CO₂
- 7. "Oven" is one component found in which of the chromatography?
 - (a) HPLC
 - (b) Size-exclusion chromatography
 - (c) Supercritical fluid chromatography
 - (d) TLC
- 8. The migration rate (v) of ion in capillary electrophoresis does not depend on :
 - (a) Electric field
 - (b) Voltage
 - (c) Length of column
 - (d) Temperature
- 9. Which metal is used as a target plate for production of X-rays in XRF spectroscopy?
 - (a) Tungsten
 - (b) Carbon
 - (c) Silicon
 - (d) Phosphorus

- 10. The transition of electron in copper from $2p^{1/2} \rightarrow 1s$ is called:
 - (a) $K\alpha_1$
 - (b) $K\alpha_2$
 - (c) $K\beta_1$
 - (d) $K\beta_2$
- 11. Which of the following is a non-destructive technique for analysis of multi-element at a time?
 - (a) Flame-atomic absorption spectrometry (Flame-AAS)
 - (b) Inductively coupled plasma-atomic emission spectroscopy (ICP-AES)
 - (c) Energy dispersive-X-ray fluorescence (ED-XRF) spectroscopy
 - (d) Inductively coupled plasma-mass spectrometry (ICP-MS)
- 12. Which of the following class of the elements can be analyzed with GF-AAS?
 - (a) Cl, Br, F
 - (b) C, N, O
 - (c) Zn, Ni, Cu
 - (d) He, Ne, Ar

- 13. The extent of the method or technique to detect the target analyte in the presence of other chemical substances is called as a:
 - (a) Sensitivity
 - (b) Selectivity
 - (c) Interference
 - (d) Robustness
- 14. Which of the following technique is best suitable for the analysis of mercury ions from soil samples?
 - (a) HG-AAS
 - (b) GF-ASS
 - (c) Flame-MS
 - (d) CV-AAS
- 15. Which of the following instruments needs a higher energy for the removal of core shell electron for surface characterization studies?
 - (a) X-ray fluorescence
 - (b) Proton induced X-ray spectroscopy
 - (c) Auger electron spectroscopy
 - (d) X-ray absorption
- 16. The compound of molecular weight > 50 KDa is analyzed with mass analyzer :
 - (a) Ion trap
 - (b) Time-of-flight
 - (c) Quadruple
 - (d) Quadruple-Quadruple

17. Which of the following ionization processes is common in LC-MS?

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- (a) APCI
- (b) CI
- (c) EI
- (d) MALDI
- 18. HG-AAS is good technique for analysis of which of the elements from water samples ?
 - (a) Li, Na, K and Ca
 - (b) Fe, Co, Cu and Zn
 - (c) As, Se, Bi and Sb
 - (d) Hg, Pb, Ni and Al
- 19. Which of the following techniques will be good for protein sequencing of COVID-19 virus?
 - (a) GC-MS
 - (b) HPLC-MS
 - (c) ICP-MS
 - (d) All techniques
- 20. Which flame is preferable for the analysis of elements like Al, Ti, Mo and V?
 - (a) Air-acetylene
 - (b) Acetylene-oxygen
 - (c) Nitrous oxide-acetylene
 - (d) Propane-acetylene

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Section—B 2 each

(Very Short Answer Type Questions)

Note: Attempt all questions in 2-3 sentences.

- 1. What are the advantages of supercritical fluid?
- 2. Give the principle of electrochromatography and write its one application.
- 3. Differentiate between ED-XRF and WD-XRF.
- 4. Give the working principle of PIXE.
- 5. Which atomic emission spectrometry is good for analysis of light elements like Li, Na, K, Ca, etc. ?
- 6. Write the name of radiation source used in atomic fluorescence spectroscopy?
- 7. Why the formation of metal hydrides is done before atomization in HG-AAS?
- 8. What is usefulness of hyphenated technique such as GC-MS and LC-MS compared to GC and HPLC?

Section—C 3 each

(Short Answer Type Questions)

Note : Attempt all questions. Write answer in ≤ 75 words.

- 1. Write the principle of ion-exchange chromatography (IC) and how the metal ions can be separated using IC.
- 2. What is electroosmotic flow? How is it useful in separation of chemical substances in capillary electrophoresis?
- 3. Give names of the detectors that can be used in PIXE.
- 4. Shortly explain the production of X-rays.
- 5. Write the principle and applications of ICP-AES.
- 6. What is the importance of selectivity and sensitivity in analytical chemistry?

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7. Give the principle of HPLC-MS for separation of compound mixture.

8. Why GF-AAS is more sensitive than Flame-AAS?

Section—D 5 each

(Long Answer Type Questions)

Note: Attempt all questions. Write answer in < 150 words.

1. Shortly describe the principle for separation of compound mixture in supercritical fluid chromatography.

Or

Briefly explain the theory and instrumentation of capillary electrophoresis.

2. What is X-ray fluorescence (XRF) spectroscopy? Comparatively explain the Energy dispersive-XRF and wavelength dispersive-XRF and their applications.

Or

Briefly describe the different components of PIXE and their applications in analysis of archaeological materials.

 Give the working principle and different application of ICP-AES.

Or

Write notes on any two of the following:

- (a) Chemical ionization in AES
- (b) Refractory compound formation in AAS
- (c) Inductively couple plasma
- 4. Write the working principle of Flame atomic absorption spectrometry and their applications.

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Write notes on the following (any two):

- (a) CV-AAS
- (b) IC-MS
- (c) HPLC-MS

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